



**OAK GROVE SCHOOL DISTRICT**

**HERMAN  
CHEMICAL HYGIENE  
PROGRAM**



## CHEMICAL HYGIENE PLAN

### INTRODUCTIONS

This document describes the Chemical Hygiene Plan for Oak Grove School District as required by OSHA's Occupational Exposure to Hazardous Chemicals in Laboratories Standard. 29CFR1910.1450.

In order to comply with the Standard, Oak Grove School District has established a plan to ensure our workplace is capable of protecting employees and students from Health Hazards associated with Hazardous Chemicals in the laboratory, and capable of keeping exposures below the Permissible Exposure Limits (PEL) specified in 29CFR1910 Subpart "Z". Questions regarding this plan should be directed to Neil Rauschhuber, Chief Operations Officer at 408-8300 ext.100357.

This plan is made readily available to the employees of Oak Grove School District, employee representatives and upon request to the Maintenance and Operations Department. The plan is located on the Oak Grove School District's internet.

A review and evaluation of this program is conducted on an annual basis by the Safety Committee and updated as necessary.

The person responsible for implementation of this plan is Neil Rauschhuber, Chief Operations Officer. The assigned Chemical Hygiene Officer is Oscar Ortiz, Director, ESD. The District's Safety Committee will review and monitor compliance.

The committee comprised of the following personnel:

- Chief Business Official
- Chief Operations Officer
- Manager of Maintenance and Operations
- Manager of Operations
- Director of Human Resources
- RTW Specialist/HR
- Director of Business
- Purchase/Finance Technician
- Teacher
- Payroll Technician
- Director of ESD

- Coordinator HR
- District Nurse
- Executive Secretary
- Executive Division Secretary
- STW
- Principal
- SIG
- AFSCME
- CSEA
- Risk Management Consultant

## **STANDARD OPERATING PROCEDURES**

For laboratory work at Oak Grove School District which involves the use of Hazardous Chemicals, standard operating procedures (SOP) have been addressed in order to reduce the potential Safety and/or Health Hazards caused by such use. These procedures include various engineering control measures such as maintenance procedures for these engineering controls including testing proper function of such equipment, the use of appropriate Personal Protective Equipment (PPE) and maintenance of such equipment.

The specific procedures implemented by Oak Grove School District are as follows:

## **LABORATORY ACTIVITIES**

Laboratory activities and demonstrations represent an essential part of science teaching. Written materials and pictures can convey an enormous amount of information, but students fully understand the concepts related to science when they participate in or observe learning activities involving laboratory experiments and demonstrations. The Science Safety Handbook for California Public Schools will be followed during the operations of the science lab.

- Science teachers should advocate safety and have the information necessary to minimize risk.
- Students have a responsibility to themselves and their classmates to learn and observe safety practices in all participatory science activities.
- Education Code Section 51202 names what should be taught in most laboratory classes, including the following practices, skills, or areas of knowledge:
  - Proper eye-safety practices
  - Proper handling of glassware and glass tubes
  - Proper setup and handling of electrical equipment
  - Safe use of chemicals in the laboratory (correct storage, handling, labeling and disposal)
  - Safe, appropriate use of heat sources in the laboratory
  - Proper prevention of exposure to bloodborne pathogens from sources such as sharp instruments, tissues and body fluids
  - First-aid procedures
  - Prompt notification to appropriate supervisors or agencies of any dangerous or potentially dangerous conditions
  - Safe, appropriate, and humane treatment of animals

- Prohibition against the use or presence of any venomous animals, poisonous plants or plant pests
- Proper fire prevention and control techniques
- Correct methods for cleanup after experiments
- Proper behavior and courtesy for laboratory situations
- Earthquake-safe behavior and evacuation routes (routes are available by the classroom door)

## **SAFE SCIENCE CLASSROOMS AND LABORATORY**

The California Code of Regulations (CCR), Title 24-Part 2, the California Building Code, Table 1004.1.2 identifies Occupant Load Factors for different types of building uses. For a science lab classroom in any kindergarten through grade 12 (K-12) educational space, the occupant load factor is 50 square feet per person. Classrooms shall not exceed the number of persons allowed.

Other design specifications related to science laboratories in the CCR, Title 5, Section 14030 (i)(1)(A-J) include the following:

Laboratories shall be designed in accordance with the planned curriculum

Size of the science laboratory is at least 1300 square feet, including storage and teacher preparation area

Science laboratory design is consistent with the requirements for proper hazardous materials management according to the California Science Safety Handbook Edition 2014

Accommodations are made for necessary safety equipment and storage of supplies (e.g., fire extinguishers, first-aid kit, master disconnect valve for gas and Safety Data Sheets (SDS)).

Secure storage areas are provided for volatile, flammable, and corrosive chemicals and cleaning agents.

Properly designated areas are provided with appropriate ventilation of hazardous materials that emit noxious fumes.

Exhaust-fume hoods, eye washes, and deluge showers are provided if applicable

Floor and ceiling ventilation is provided in areas where chemicals are stored

Room is provided for movement of students around fixed-learning stations

## **CHEMICAL INFORMATION**

The sites will ensure that hazardous materials are properly used and stored.

The sites will follow the CDE, in cooperation with the Division of Occupational Safety and Health, the list of chemicals used in a school program. This list includes the potential hazards and estimated shelf life of each chemical and chemical compound. The District will conduct regular removal and disposal of all chemicals whose estimated shelf life has elapsed through an Environmental company.

The Chemical Hygiene Plan shall be readily available to employees, employee representatives, and, on request, the Chief of the Division of Occupational Safety and Health. The plan is posted on the District's website.

### **SAFETY PRECAUTIONS FOR SCHOOL SCIENCE LABORATORY**

Employee information and training on the hazards of chemicals present in the work area shall be provided at the time of an employee's initial assignment to his or her work area and prior to assignments involving new exposure situations. Refresher information and training shall be provided at intervals determined by the District.

The California Science Safety Handbook for California Public Schools Edition 2014 is readily available to employees and employee representatives. The plan should be followed at all times. The plan is posted on the District's website.

Bloodborne Pathogens (CCR, Title 8, General Industry Safety Orders, Section 5193) applies to all employees that are subject to reasonably anticipated exposure of their skin, eyes or mucous membranes, or through parenteral contact, to blood or other potentially infectious materials as a result of the performance of the duties. The Bloodborne Pathogen Plan is included in the Injury and Illness Prevention Plan (IIPP) and is posted on the District's website. A hard copy is available in the site front office.

Because science classes include a variety of hands-on laboratory activities in which the use of glassware and sharp instruments may result in cuts and abrasions, science teachers should be an integral part of, and comply with their school's exposure control plan. Science teachers will be trained annually as required by the IIPP.

The science teachers should be fully acquainted with first-aid procedures.

Understand the potential hazards of all the materials, processes, and equipment that will be used in the school lab.

Report any student injury or accident immediately on the school district's accident report form.

The use of approved eye-protection devices is required of all persons performing science activities that involve potential hazards to the eyes.

An eyewash station must be available in each laboratory/classroom.

Chemicals should be stored according to their compatibility group in a single safe and practical storage pattern.

Chemicals should be stored in chemically inert containers appropriate for the type and quantity of chemical.

Chemicals should be stored in a separate, locked area away from the classroom, if possible. A NFPA warning symbol should be placed on the front of the cabinet showing the highest hazard rating in each category of any chemical stored in the cabinet.

Regularly inspect chemicals and other supplies. Annually update the chemical inventory and discard any leaking, damaged, empty or unlabeled containers according to protocol.

Maintain a copy of the inventory for local emergency responders.

Maintain a complete inventory of every chemical in storage. Keep copies of the inventory and all SDS in the storage area. The chemical inventory should be reviewed and revised annually.

Chemicals should be purchased in a quantity sufficient to be used within one year. The date of receipt should be written on the label.

When labeling chemicals that have been removed from the original container, give the chemical name or trade name of the product on the label. The label must include the name of the chemical manufacturer along with the address and telephone number. Indicate the strength of the chemical, especially if it was prepared on site. Date all bottles.

Food for human consumption should not be stored in refrigerators or cabinets or on shelves used for storing chemicals or biological materials. Food should not be eaten in science laboratories or storage areas because of the danger of ingesting toxic or corrosive substances.

In an experiment or demonstration involving any flammable liquid (such as alcohol), care must be taken that any flame in the room is at an absolutely safe distance from the volatile liquid.

Teachers and students should be familiar with the operation of all fire extinguishers in the laboratory.

Teachers should be familiar with the location of all master controls for utilities, especially the master valve in each room for the gas outlets.

Good housekeeping is essential in the laboratory.

Laboratories should be locked when not in use.

The custodial staff should be alerted to general hazards they may encounter.

Safety Data Sheets (SDS) for new chemicals must be submitted to Fred Dickey, Manager of Transportation and Operations. The chemical will be logged on the on-line program.

## **EVACUATION**

The school's evacuation plans, hazardous materials procedures are available to all employees. Important local telephone numbers are available in case of an emergency. The phone numbers are posted at the classroom doors.

## **KNOWLEDGE**

It is the responsibility of all science laboratory staff to have the knowledge and understanding of:

- Science Safety Handbook for California Public Schools Edition 2014
- Oak Grove School District's Chemical Hygiene Plan
- Oak Grove School District's Hazardous Waste Disposal plan

- Oak Grove School District's Hazardous Communication plan
- Oak Grove School District's Fire Prevention Plan
- Oak Grove School District's IIPP (including but not limited to Bloodborne Pathogens)
- Use of Personal Protective Equipment
- Safety Data Sheet (SDS) procedure
- School Eye Safety: Sections 32030, 32031 and 32033 of the EC
- Earthquake Emergency Procedures as per ED 322829a0(B)(ii), 35295, 35296,35297, 40041.5
- Personal Protective Equipment
- Use of Animals in Public Instruction: EC Section 51540
- Know the storage, handling and safety requirements for each chemical used.

### **STEPS FOR ESTABLISHING A SAFE CHEMICAL STORAGE AREA**

Safe storage and use of chemicals and the supervision of laboratory safety are addressed in several sections of the EC and the CCR. Title 8. The following is a summary of some of the relevant stipulations in those codes:

- Each school that offers laboratory work should designate a trained member of the professional staff as the person “responsible for the review, updating, and carrying out of the school’s adopted procedures for laboratory safety”. (EC Section 49341[b])
- Ensure hazardous material are properly used and stored EC Section 49401.5[a])
- Oak Grove School District’s procedures will be followed for the regular removal and disposal of all chemicals whose estimated shelf life has elapsed (EC section 4911[b]).
- The District’s Hazardous Communication Program for employees will be followed. Laboratories must maintain labels and SDS of incoming hazardous substances and ensure that they are readily available (CCR, Title 8, Section 5194[b])
- Chemicals should not be stored above eye level.
- Chemicals should not be stored under sinks; with the exception of household bleach and cleaners.
- Solids should be stored higher than liquids.
- No chemicals should be stored on the floor.
- Earthquake lips or barriers are in place on open storage shelves.
- Storage cabinets for corrosive and flammable chemicals are used appropriately.
- The storeroom door is self-closing and is locked.
- Compressed gas cylinders are secured upright to the wall, with caps in place. Flammable gases are separated from oxidizing gases by a one-hour fire wall or at least 25 feet.
- Staff members are adequately trained in chemical storage policies.
- Disposal of chemicals will be processed through the M&O Department, Neil Rauschhuber. Disposal of hazardous chemicals will be through a licensed contracting firm. A manifest will be maintained for three years.

### **CHEMICAL SEGREGATION**

The storage area should be to separate materials according to chemical specific need. Use the following broad hazard classes as a guide:

Store like materials with like. It is essential to segregate incompatible substances to prevent dangerous interactions. All newly purchased chemicals should have a label on them identifying their hazard category (e.g. flammable, corrosive, oxidising, toxic etc.). A list of commonly used chemicals that should be segregated is listed below to assist storage.

Store the minimum stock levels of hazardous chemicals in the laboratory

Dispose of hazardous chemicals that are no longer required

Store large breakable containers, particularly of liquids, below shoulder height

Ensure containers and bottle tops are sealed properly to avoid unnecessary leakage of fumes / vapours

### **Storage of specific chemicals**

- Flammable Solvents - e.g. alcohols, toluene, hexane

The vapour above the liquid of these solvents represents the main source of danger from flammable liquids. This vapour is very susceptible to ignition by naked flames, sparks from electric switches (e.g. thermostats) electric motors or from sparks produced electrostatically by friction. Precautions must therefore be taken to prevent contact between any of these and concentrated vapours of flammable liquids.

Flammable solvents should be stored in specialised metal flammable solvent containers (cabinets), clearly labelled and positioned away from doors or other means of escape from the laboratory. It should be noted that no more than 50 litres of flammable material may be kept in any one laboratory room to reduce the risk of a serious laboratory fire. Working volumes of flammable solvents (i.e. those kept on the bench) should not exceed 500 ml (this volume can be easily contained should an accident/fire occur) and the solvent must be kept in a suitable closed vessel.

Flammable solvents should not be stored in fume hoods or vented cabinets, since the airflow will fan any fire and may also spread the fire to other parts of the building via the ventilation ducting.

Flammable solvents must never be stored in a refrigerator unless they are known to be spark proof. It is now departmental policy that all laboratory fridges are spark proofed to avoid the possibility of an internal light or thermostat control unit providing a source of ignition for vapours produced from flammable substances.

Flammable solvents must never be stored with oxidising agents (e.g. sodium [hypochlorite](#) – bleach, [iodine](#) and other halogens, hydrogen peroxide, nitric acid, potassium permanganate etc.) reducing agents (e.g. [sodium borohydride](#), [lithium aluminum hydride](#)...etc) or concentrated acids (e.g. concentrated sulphuric or hydrochloric acids).

A list of commonly used chemicals which should be segregated is listed below to assist storage. The SDS should always be consulted if further information is required on storage and chemical incompatibilities of a particular chemical substance.

- Chlorinated solvents (e.g. chloroform, dichloromethane (DCM) trichloroethylene)



Chlorinated solvents are best stored in ventilated cabinets separately from flammable (non-chlorinated) solvents because violent reactions can result from the mixing of certain flammable and chlorinated solvents and toxic gases such as phosgene (as well as hydrogen chloride and chlorine) can be produced. They should not be stored with alkali metals such as lithium, potassium or sodium, since any mixing can cause an explosion. They can be stored in metal containers if ventilated storage is not available.

- Acids and Alkalis

Acids and alkalis are both corrosive substances. They must be stored separately since any accidental mixing of concentrated materials will generate large quantities of heat and fumes. They can be stored in a vented or metal cabinet so long as they are in a containment tray to prevent any spillages.

Consideration must be given to the effects of corrosive fumes on any metal in the fittings and construction of the container. The use of ventilated cabinets are recommended where possible, allowing the removal of fumes at source. All containers / bottle tops must be suitably sealed to avoid unnecessary leakage of fumes (NB parafilm can be used to seal container lids but should be checked/replaced at regular intervals).

- Oxidisers (e.g. peroxides, perchlorates and nitrates)

Oxidising substances should be stored in a metal cabinet and away from organic matter such as wood and paper (NB oxidising agents should never be stored in a wooden cabinet!). Oxidising agents must never be stored with flammable solvents or reducing agents since fires and explosions can result after any spillage, even without a naked flame or heat.

Perchloric acid is an extremely strong oxidising agent (especially in the concentrated form), which can react explosively with organic materials. It should ideally be stored separately on a metal tray of sand within a cabinet, away from organic materials or dehydrating agents such as sulphuric acid.

### **Incompatibility of Common Laboratory Chemicals**

The improper storage or mixing of chemicals can result in serious accidents and even disasters. Violent reactions could occur due to the storing or mixing incompatible chemicals. The following is a list of some incompatible common laboratory chemicals. Before storing or mixing any chemicals, consult this list or the chemicals' MSDS. This is only a partial list that includes some of the more common academic laboratory chemicals. Please note that the absence of a chemical from the list does not mean that it is necessarily safe to mix it with any other chemical! You should always check with the SDS if in doubt.

<b>Chemical</b>	<b>Incompatible with</b>
Acetic acid	Chromic acid, nitric acid, hydroxyl compounds, ethylene glycol, perchloric acid, peroxides, permanganates
Acetylene	Chlorine, bromine, copper, fluorine, silver, mercury

Acetone	Concentrated nitric acid and sulphuric acid mixtures
Alkali and alkaline earth metals	Water, carbon tetrachloride or other chlorinated hydrocarbons, carbon dioxide, halogens
Ammonia (anhydrous)	Mercury(e.g., in manometers), chlorine, calcium hypochlorite, iodine, bromine, hydrofluoric acid (anhydrous)
Ammonium nitrate	Acids, powdered metals, flammable liquids, chlorates, nitrites, sulphur, finely divided organic combustible materials
Aniline	Nitric acid, hydrogen peroxide
Arsenical materials	Any reducing agent
Azides	Acids
Bromine	See chlorine
Calcium oxide	Water
Carbon (activated)	Calcium hypochlorite, all oxidizing agents
Chlorates	Ammonium salts, acids, powdered metals, sulphur, finely divided organic or combustible materials
Chromic acid and chromium trioxide	Acetic acid, naphthalene, camphor, glycerol. Alcohol, flammable liquids in general
Chlorine	Ammonia, acetylene, butadiene, butane, methane, propane (or other petroleum gases), hydrogen, sodium carbide, benzene, finely divided metals, turpentine
Chlorine dioxide	Ammonia, methane, phosphine, hydrogen sulphide
Copper	Acetylene, hydrogen peroxide
Cumene hydroperoxide	Acids (organic and inorganic)

Cyanides	acids
Flammable liquids	Ammonium nitrate, chromic acid, hydrogen peroxide, nitric acid, sodium peroxide, halogens
Fluorine	All other chemicals
Hydrocarbons (such as butane, propane, benzene)	Fluorine, chlorine, bromine, chromic acid, sodium peroxide
Hydrocyanic acid	Nitric acid, alkali
Hydrofluoric acid (anhydrous)	Ammonia (aqueous or anhydrous)
Hydrogen sulphide	Fuming nitric acid, oxidizing gases
Hypochlorites	Acids, activated carbon
Iodine	Acetylene, ammonia (aqueous or anhydrous), hydrogen
Mercury	Acetylene, fulminic acid, ammonia
Nitrates	Acids
Nitric acid (concentrated)	Acetic acid, aniline, chromic acid, hydrocyanic acid, hydrogen sulphide, flammable liquids and gases, copper, brass, any heavy metals
Nitrites	Acids
Nitroparaffins	Inorganic bases, amines
Oxalic acid	Silver, mercury
Oxygen	Oils, grease, hydrogen, flammable liquids, solids, and gases

Perchloric acid	Acetic acid, anhydride, bismuth and its alloys, alcohols, paper, wood, grease, oils
Peroxides, organic	Acids (organic or mineral), avoid friction, store cold
Phosphorus (white)	Air, oxygen, alkalies, reducing agents
Potassium chlorate	Sulphuric and other acids
Potassium perchlorate (see also chlorates)	Sulphuric and other acids
Potassium permanganate	Glycerol, ethylene glycol, benzaldehyde, sulphuric acid
Selenides	Reducing agents
Silver	Acetylene, oxalic acid, tartaric acid, ammonium compounds, fulminic acid
Sodium	Carbon tetrachloride, carbon dioxide, water
Sodium nitrite	Ammonium nitrate and other ammonium salts
Sodium peroxide	Ethyl and methyl alcohol, glacial acetic acid, acetic anhydride, benzaldehyde, carbon disulfide, glycerin, ethylene glycol, ethyl acetate, methyl acetate, furfural
Sulphides	Acids
Sulphuric acid	Potassium chlorate, potassium perchlorate, potassium permanganate (similar compounds of light metal, such as sodium, lithium)
Tellurides	Reducing agents

## **EMPLOYEE EXPOSURE**

Oak Grove School District makes an initial determination of employee's exposure to any substance regulated by OSHA Standard (29CFR1910.1001 through 20CFR1910.1050) which requires monitoring if there is reason to believe exposure levels for substances routinely exceed

the Action Level or in the absence of an Action Level, the Permissible Exposure Limit (PEL). If initial monitoring discloses employee exposure over the Action Level and/or Permissible Exposure Limit, exposure monitoring and cessation of monitoring shall be in accordance with the relevant OSHA Standard.

Where the use of respirators is determined to be necessary to maintain exposure below permissible limits, appropriate respirator protection shall be provided at NO COST to employees. Respirators shall be selected and used in accordance with requirements in 29CFR1910.134(b)(d)(e) & (f).

Oak Grove School District notifies employees of these results in writing either individually or by posting results in the staff lounge within fifteen (15) working days after receipt of monitoring results.

### **EMPLOYEE INFORMATION & TRAINING REQUIREMENTS**

Oak Grove School District provides employees with information and training on hazardous chemicals in their work area, at time of initial assignment, and prior to assignments involving new exposure situations to ensure employees are apprised of the hazards of chemicals present in their work area. Initial information and training sessions for new employees are conducted by Human Resources.

The following information is conveyed to employees:

- The contents of the Occupational Exposure to Hazardous Chemicals in Laboratories Standard 29CFR1910.1450 and its appendices.
- The location and availability of the Chemical Hygiene Plan.
- The signs and symptoms associated with exposure to hazardous chemicals used in the laboratory
- The location and availability of known reference material on the hazards, safe handling, storage & disposal of hazardous chemicals found in the laboratory including but not limited to Safety Data Sheets (SDS) received from chemical suppliers.

In addition to above information, employees are trained on the following:

- Methods and observations which may be used to detect the presence or release of a hazardous chemical in the work area.
- The physical and health hazards of the chemicals.
- The measures employees can take to protect themselves from these hazards, including specific procedures which have been implemented to protect employees from exposure to hazardous chemicals, such as appropriate work practices, emergency procedures, and personal protective equipment to be used.
- The applicable details of the Chemical Hygiene Plan.
- IIPP, including Bloodborne Pathogens

## **HAZARD IDENTIFICATION**

With respect to labels and Safety Data Sheets (SDS), Oak Grove School District assures labels on incoming containers of hazardous chemicals are not removed or defaced and Safety Data Sheets that are received with incoming shipments of hazardous chemicals are entered in the SDS 3E system on-line and readily accessible to laboratory employees. A SDS poster giving instructions on receiving a safety data sheet from the database is available in the science lab.

For chemical substances developed in the laboratory, the following procedures have been implemented for the following circumstances:

- For known composition for chemical substances produce exclusively for laboratory use which are determined to be hazardous, the appropriate training is provided
- For unknown composition of chemicals produced as a by-product, the substance is assumed to be hazardous and the appropriate standard procedures are implemented
- For chemical substances produced for another user outside of the laboratory, Oak Grove School District complies with the Hazard Communication Standard 29CFR1910.1200, including the requirements of preparation of SDS and labeling.

## **EMPLOYEE EXPOSURE**

Oak Grove School District makes an initial determination of employee's exposure to any substance regulated by OSHA Standard (29CFR1910.1001 through 20CFR1910.1050) which requires monitoring if there is reason to believe exposure levels for substances routinely exceed the Action Level or in the absence of an Action Level, the Permissible Exposure Limit (PEL). If initial monitoring discloses employee exposure over the Action Level and/or Permissible Exposure Limit, exposure monitoring and cessation of monitoring shall be in accordance with the relevant OSHA Standard.

Where the use of respirators is determined to be necessary to maintain exposure below permissible limits, appropriate respirator protection shall be provided at NO COST to employees. Respirators shall be selected and used in accordance with requirements in 29CFR1910.134(b)(d)(e) & (f).

Oak Grove School District notifies employees of these results in writing either individually or by posting results in the staff lounge within fifteen (15) working days after receipt of monitoring results.

## **MEDICAL PROGRAM**

Oak Grove School District provides all employees who work with hazardous chemicals an opportunity to receive medical attention, including any follow-up examinations which the examining physician determines to be necessary, for the following circumstances:

- Whenever an employee develops signs or symptoms associated with a hazardous chemical to which the employee may have been exposed in the laboratory, the employee is provided an opportunity to receive an appropriate medical examination.

- Where exposure monitoring reveals an exposure routinely above the Action Level, or in the absence of an Action Level above the Permissible Exposure Limit (PEL), for an OSHA regulated substance for which there are exposure monitoring and medical surveillance requirements. Medical Surveillance shall be established for the effected employee as prescribed by the particular OSHA Standard. (Medical Surveillance & Exposure requirements as per 29CFR1910.1001 through 20CFR1910.1050).
- Whenever an event takes place in the work area such as a spill, leak, explosion or occurrence resulting in the likelihood of a hazardous chemical exposure, the effected employee(s) is provided an opportunity for a medical consultation. This consultation is for the purpose of determining the need for a medical examination. An appropriate medical examination is provided as necessary through the Injury and Illness Prevention Plan (IIPP).

Medical examinations are provided by the District's Workers Compensation medical provider at no cost to the employee without loss of pay and at a reasonable time and place. Employees should report to Karen Lemm, Workers Compensation Specialist, any injury or illness immediately.

The physician shall be provided with the following:

- The identity of hazardous chemical(s) to which the employee may have been exposed
- A description of the conditions under which the exposure occurred including quantitative exposure data, if available
- A description of the signs and symptoms of exposure the employee is experiencing if any.

The Safety Data Sheets (SDS) are on-line through the 3E company. An employee has 24/7 access to the SDS at **1-800-451-8346**.

#### **HOW TO REQUEST A SDS:**

To request a safety data sheet (SDS), an authorized employee simply calls 1-800-451-8346 or mail the request to 3E Company. To ensure your request is quickly processed, it is extremely important to provide as much of the following product information as possible:

- Produce Name
- Manufacturer name
- Product number (found on side of container)
- UPC code (if available)

Please be as specific as possible when requesting SDS for a product. Separate SDS are maintained for products that have even very minor differences from others.

A written opinion is obtained from the examining physician for an examination or consultation which is required by the Standard. The opinion shall include:

- Any recommendation for further medical follow-up
- The results of the medical examination and any associated tests
- Any medical condition that may be revealed in the course of the examination which may place the employee at increased risk as a result of exposure to a hazardous chemical found in the workplace
- A statement the employee has been informed by the physician of the results of the consultation or medical examination and any medical condition that may require further examination or treatment.
- The written opinion shall not include any findings/diagnoses which is not related to an occupational exposure.

## **RECORDKEEPING**

Oak Grove School District establishes and maintains for each employee an accurate record of any measurements taken to monitor employee exposure and any medical consultation and examination including tests or written opinions as required by the Standard. These records are kept, transferred, and made available in accordance with 29CFR1910.20, Access to Employee Exposure and Medical Records.

## **INFORMATION AND RESOURCES**

**SDS** are filed on-line and available to all employees.

Chemical inventory is available on-line and at the M&O Office.

The Chemical Hygiene Plan is reviewed and approved by the Safety Committee and is available on-line.

Oak Grove School District OSHA required policies and procedures are available on-line.

The Chemical Hygiene Plan was approved by the Safety Committee on April 21, 2015

The plan is revised annually and was last updated on June 30, 2017..



## HERMAN MIDDLE SCHOOL CHEMICAL INVENTORY JUNE 2017

Location	Product Name	Mfr.	Mfr. Part #	Physical State	Mfr. City	Mfr. State	Container Type	Container units	Unit of measure for size of container		Notes
Room 25											
Flammable cabinet 009870	Calcium Acetate	Flinn Sci.	AP6651A	Liquid	Batavia	IL	Plastic	1	500 ml		
Cabinet 009870	Potassium Iodate	Flinn	AP4601A	Liquid	Batavia	IL	Plastic	3	800 ml		
Cabinet 009870	Sodium Metabisulfite	Flinn	AP4601B	Liquid	Batavia	IL	Plastic	1	4 gr		
Cabinet 009870	Hydrogen Peroxide 30%	Flinn	AP2092A	Liquid	Batavia	IL	Plastic	1	50 ml		
Cabinet 009870	Phenolphthalein	Flinn	P0018	Powder	Batavia	IL	Plastic	1	50 grams		
Cabinet 009870	Sodium Chloride	Flinn	AP4865B	Crystal	Batavia	IL	Plastic	1	50 grams		
Cabinet 009870	Strontium Chloride	Flinn	AP5607F	Crystal	Batavia	IL	Plastic	1	50 grams		
Cabinet 009870	Potassium Chloride	Flinn	AP5607D	Powder	Batavia	IL	Plastic	1	50 grams		
Cabinet 009870	Lithium Chloride	Flinn	AP5607C	Powder	Batavia	IL	Plastic	1	50 grams		
Cabinet 009870	Methylene Blue Solution	Flinn	AP8653C	Liquid	Batavia	IL	Plastic	1	10 ml		

Cabinet 009870	Dextrose	Flinn	AP8653B	Liquid	Batavia	IL	Plastic	1	500 ml		
Cabinet 009870	Sodium Bicarbonate	Home Scientist	x	Tablets			Plastic	1	15 g		
Cabinet 009870	Phenolphthalein .5%	Home Scientist	x	Liquid			Plastic	1	15 ml		
Cabinet 009870	Iron	Home Scientist	x	Solid			Plastic	1	nail		
Cabinet 009870	Magnesium	Home Scientist	x	Solid			Plastic	1	strip		
Cabinet 009870	Copper wire	Home Scientist	x	Solid				11			
Cabinet 009870	Potassium Ferricyanide .1M	Home Scientist	X	Liquid			Plastic	1	15 ml		
Cabinet 009870	Iron Chloride .1m	Home Scientist	X	Liquid			Plastic	1	15 ml		
Cabinet 009870	Iron Sulfate .1 m	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Phosphoric acid .1M	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Sodium Carbonate 1M	Home Scientist	X	Liquid				1	10 ml		
Cabinet 009870	Ammonia 6 m	Home Scientist	X	Liquid				1	10 ml		
Cabinet 009870	Sodium Salicylate 200/m	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Sodium Sulfide	Home Scientist	X	Liquid				1	10 ML		

Cabinet 009870	Potassium Dicchromate .1M	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	N-Butanol	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Vegetable oil	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Starch indicator	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Ascorbic Acid	Home Scientist	X	Tablets				18	9 g		
Cabinet 009870	Salicylic Acid	Home Scientist	X	Powder				1	10 g		
Cabinet 009870	Thymol blue	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Sodium Hydroxide 6M	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Methyl Orange .1 %	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Methyl Red .02%	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Bromothymol Blue .1%	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Sodium Ferricyanide .1M	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Sodium Borate	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Turmeric	Home Scientist	X	Liquid				1	5 ml		

Cabinet 009870	Sodium Bi-Sulfite	Home Scientist	X	Liquid				1	10 ml		
Cabinet 009870	Sodium Thiosulfate 1 M	Home Scientist	X	Liquid				1	15 ml		
Cabinet 009870	Activated charcoal	Home Scientist	X	Powder				1	10 g		
Cabinet 009870	Magnesium Sulfate	Home Scientist	X	Crystal				1	15 g		
Cabinet 009870	Povidone Iodine	Walgreens		Liquid	Deerfield	IL	Plastic	1	8 Oz		
Cabinet 009871	Glycerin	Whitworth		Liquid	Gardena	CA	Glass	1	4 oz.		
Cabinet 009871	Glycerin	Humco		Liquid	Texarcana	Texas	Plastic	1	4 oz		
Cabinet 009871	Methyl Orange	Frey Scientific	579402	Liquid	Mansfield	OH	Plastic	3	100 ml		
Cabinet 009871	Metholene Blue 1%	Fischer	S71325	Liquid	Chicago	IL	Plastic	1	100 ml		
Cabinet 009871	Methylene Blue	Science Shop	C5400	Liquid	San Jose	CA	Plastic	1	100 ml		
Cabinet 009871	Methylene Blue	Frey Scientific	76091	Liquid	Mansfield	OH	Plastic	1	500 ml		
Cabinet 009871	Bromthymol Blue .045%	Frey Scientific	71831	Liquid	Mansfield	OH	Plastic	2	500 ml		
Cabinet 009871	Guar Gum	Flinn Sci	G0038	Powder	Batavia	IL	Plastic	1	25 g		
Cabinet 009871	Bio Salts	Science kit and Boreal Labs	80752-02	Powder	St. catharines	Ontario	Plastic	1	100 g		

Cabinet 009871	Bio Salts	Flinn Sci	B0133	Powder	Batavia	IL	Plastic	1	25 g		
Cabinet 009871	Glycerin	Whitworth		Liquid	Gardena	CA	Glass	1	2 oz		
Cabinet 009871	Phenolphthalein	Nasco	SA9652M	Powder	Modesto	CA	Plastic	1	100 g		
Cabinet 009871	Nutrient Agar	Flinn Sci	N0019	Powder	Batavia	IL	Plastic	1	100 g		
Cabinet 009871	Agar	Flinn Sci	A00112	P	Batavia	IL	Plastic	1	100 g		
Cabinet 009871	Mineral Oil	Flinn Sci	M0065	Liquid	Batavia	IL	Plastic	1	4 L		
Cabinet 009871	Guar Gum	Flinn Sci	G0040	Powder	Batavia	IL	Plastic	1	100 g		
Cabinet 009871	Starch	Flinn Sci	S0282	Powder	Batavia	IL	Plastic	2	2 kg		
Cabinet 009871	Isoprophyl Alcohol 70%	Good Sense		Liquid	Peach Tree	Georgia	Plastic	1	8 Oz		Moved to flammable cabinet
Cabinet 009871	Isoprophyl Alcohol 70%	Up & Up		Liquid	Minneapolis	MN	Plastic	4	32 oz		Moved to flammable cabinet
Under Sink	Vinegar	Signature Kitchens		Liquid	Pleasanton	CA	Plastic	1	32 oz		
Under Sink	Ammonia	Raleys	K11207	Liquid	Sacramento	CA	Plastic	1	20 oz		
Under Sink	Boraxo	20 Mule Team		Powder	Scottsdale	AZ	Cardboard	1	5 lbs		
Under Sink	Pepto Bismol	Proctor & Gamble		Liquid	Cincinnati	OH	Plastic	1	16 Oz		
Under Sink	Mineral Oil	Safeway		Liquid	Pleasanton	CA	Plastic	1	5 oz		

Storage closet between 25 and 28	Zinc	Flinn	Z0004	Solid chunks	Batavia	IL	Plastic	1	2 kilograms		Flammable: move to flammable cabinet
Storage closet between 25 and 28	Zinc	Flinn	Z0003	Solid chunks	Batavia	IL	Plastic	1	500 grams		Flammable: move to flammable cabinet
Storage closet between 25 and 28	Iron	Flinn	10012	Filings	Batavia	IL	Plastic	2	2 kilograms		
Storage closet between 25 and 28	Aluminium Shot	Flinn	A0262	Shot	Batavia	IL	Plastic	1	100 gr		
Storage closet between 25 and 28	Zinc metal	Nasco	17059M	Solid chunks	Modesto	CA	Plastic	1	500 grams		Flammable: move to flammable cabinet
Storage closet between 25 and 28	Potassium Chromate	Frey Sci	77249		Mansfield	OH	Plastic	1	15 grams		Oxidizer: should be stored separate
Storage closet between 25 and 28	Potassium Chromate	Fischer	571231	Crystal	Chicago	IL	Plastic	1	500 grams		Oxidizer: should be stored separate
Storage closet between 25 and 28	Potassium Chromate	Frey Sci	77190	Granular	Mansfield	OH	Plastic	1	500 grams		Oxidizer: should be stored separate

n 25 and 28											
Storage closet between n 25 and 28	Black Iron Oxide	Alpha Chemical		Powder	Stoughton	MA	Plastic bag	1	1 lb		
Storage closet between n 25 and 28	Sucrose	Nasco	SA9772 M	Granule	Fort Atkinson	WI	Plastic	1	2 kilogram		
Storage closet between n 25 and 28	Sodium Chloride	Nasco	SB7890 M	Crystal	Fort Atkinson	WI	Plastic	1	2 kilograms		
Storage closet between n 25 and 28	Copper Metal Powder	Frey	S70077-1	Powder	Mansfield	OH	Plastic	2	500 grams		
Storage closet between n 25 and 28	Potassium Chromate Crystals	Fisher	S71231	Crystals	Chicago	ILL	Plastic	1	500 grams		Oxidizer. Should be stored separate.
Storage closet between n 25 and 28	Silver Nitrate	Anachemica Chemicals		Liquid	Rouses Point	NY	Plastic	1	3 oz		
Storage closet between n 25 and 28	Hydrogen Peroxide 6%	Flinn	H0029	Liquid	Batavia	IL	Plastic	1	1 gallon		
Storage closet between	Zinc Sulfate	Nasco	SB7901 M	Crystals	Fort Atkinson	WI	Plastic	1	100 grams		

n 25 and 28											
Storage closet between n 25 and 28	Magnesium Sulfate	Flinn	M0016	Crystals	Batavia	IL	Plastic	1	100 grams		
Storage closet between n 25 and 28	Potassium Sulfate	Fisher	S80147-1	Powder	Chicago	IL	Plastic	1	500 grams		
Storage closet between n 25 and 28	Polyvinyl alcohol 4%	Flinn	P0209	Liquid	Batavia	IL	Plastic	1	500 ml		
Corrosive Cabinet	Sodium Hydroxide .1m	Nasco	SB17032 M	Liquid	Fort Atkinson	WI	Plastic	1	500 ml		
Corrosive Cabinet	Sodium Hydroxide 2 m	Flinn	AP88321	Liquid	Batavia	IL	Plastic1		500 ml		
Corrosive Cabinet	Sodium Hydroxide	Fischer	S71993-2	Liquid	Chicago	IL	Plastic	2	1 liter		
Corrosive Cabinet	Hydrogen Peroxide	Flinn	Hoo8	Liquid	Batavia	IL	Plastic	1	500 ML		
Storage closet between n 25 and 28	Ammonia	Safeway			Pleasanton	CA	Plastic	1	8 oz		Dated for 6/17 removal
Storage closet between	Cascade Dishwasher Liquid	Procter and Gamble		Liquid	Cincinnati	OH	Plastic	1	120 oz		



n 25 and 28											
Storage closet between n 25 and 28	Stayflow Starch	Dial		Liquid	Scottsdale	AZ	Plastic	1	2 quarts		
Storage closet between n 25 and 28	Aspirin	Top Care		Tablet			Plastic	1	300 tabs		
Storage closet between n 25 and 28	Milk of Magnesia			Liquid			Plastic	3	20 oz		
Storage closet between n 25 and 28	Windex	Johnson and Family		Liquid	New Brunswick	NJ	Plastic	1	5 oz.		
Corrosive Cabinet 9869	Hydrogen Peroxide 30%	Flinn	H0008	Liquid	Batavia	IL	Plastic	1	500 ml		
Storage room 25	A-9 Aluminum Cutting Fluid	Relton Corp		Liquid	Arcadia	CA	Metal	1	16 oz		
Storage room 25	Simulated Blood	Flinn	FB1225E	Liquid	Batavia	IL	Plastic	2	30 ml		
Storage room 25	Simulated Anti RH sera	Flinn	FB1225A	Liquid	Batavia	IL	Plastic	2	30 ml		
Storage room 25	Simulated Blood	Flinn	FB1225D	Liquid	Batavia	IL	Plastic	2	30 ml		
Storage room 25	Simulated Anti B Sera	Flinn	FB 1225C	Liquid	Batavia	IL	Plastic	2	30 ml		

Storage room 25	Simulated Anti A Sera	Flinn	FB1225B	Liquid	Batavia	IL	Plastic	2	30ml		
Room 29	1.0 m Oxalic Acid	GSE		Liquid			Plastic	11	3 oz		Date for removal 6/17
Room 29	Stewarts Blue	Luther Ford		Liquid	Bloomington	MN	Plastic	1	8 oz		
Room 29	Cascade	Proctor and Gamble		Liquid	Cincinnati	OH	Plastic1	1	45 oz		

Location	Product Name	Mfr.	Mfr. Part #	Physical State	Mfr. City	Mfr. State	Mfr. Phone	Container Type	Container units	Unit of measure for size of container	Notes
Rm 108	Vinegar	Four Monks	002129	Liquid	Mt. Prospect	IL		Plastic	2	Gallon	
Storage rm between 108 and 110	Sodium Hydroxide	Mallinckrodt	7680	Pellets	St Louis	MO		Plastic	2	10 lbs	
Storage rm between 108 and 110	Pepsin	Science kit and Boreal Laboratories	85035	Powder	San Opispo	CA		Plastic	1	25 g	
Storage rm between 108 and 110	Suave Hair Spray	Suave		Spray	Turnbull	CT		Metal Can	1	10 oz	
	Dawn Soap	Dawn		Liquid	Cincinnati	OH		Plastic	1	300 ml	
	CLR	Jelmar		Liquid	Lincoln	IL		Plastic	1	14 oz	
	Goo Gone	Magic		Liquid	Bedford	OH		Plastic	1	12 oz	
	Zinc Metal	Central Scientific	39190-01P	Chunks	Chicago	IL		Cardboard	1	1 lb	
	D-Glucose	Science Kit&Boreal	81660-03	Powder	San Opispo	CA		Plastic	1	500 g	
	Sodium Iodine	Carolina	86-9710	Powder	Burlington	NC		Plastic	1	100 g	
	Iron filings	Science Kit & Boreal	21944-10	Filings	San Opispo	CA		Cardboard	2	8 Oz	
	Iron filings	Merrell Scientific		Filings				Plastic	1	1 lb	
	Sodium Bicarbonate	Drake Brothers		Powder	Menomonie falls	WI		Plastic	1	1 lb	

	Sodium Carbonate	Science Kit & Boreal		Powder	San Opispo	CA		Plastic	1	500 g	
	Sodium Chloride	Science Kit & Boreal	86795-03	Powder	San Opispo	CA		Plastic	2	500 g	
	Potassium Chloride	Mallinckrodt		Powder	St Louis	MO		Plastic	1	1 lb	
	Lithium Chloride	Mallinckrodt		Powder	St Louis	MO		Plastic	1	1 lb	
	Potassium Carbonate	Carolina		Powder	Burlington	NC		Plastic	1	500 g	
	Sodium Hydroxide	LAS-Aids		Liquid	Bohemia	NY		Plastic			
	Sodium Tetraborate	Unknown		Powder				Plastic	1	100 g	
	Crystal Violet	Lab Aids		Liquid	Farmingdale	NY		Plastic	1	1 oz	
	L-Ascorbic Acid	Science Kit & Boral	80570-02	Powder	San Opispo	CA		Plastic	1	25 g	
	Bromothymol Blue	Nasco	SA9401	Liquid	Salida	CA		Plastic	1	250 ml	
	Glucose Solution	Lab Aids		Liquid	Farmingdale	NY		Plastic	3		
	Thiosulfate	LaMotte	4169	Liquid	Chestertown	MD		Plastic	1	60 ml	
	Mang Sulf	LaMotte		Liquid	Chestertown	MD		Plastic	1	60 ml	Dated 6/17
	Sulfamic Acid	LaMotte	6286H	Liquid	Chestertown	MD		Plastic	1	50g	
	Ammonia	Rite Aid		Liquid	Camp Hill	PA		Plastic	1	2 qt	
	Supreme Clean	Raleys		Liquid	West Sacramento	CA		Plastic	1	64 oz	
	White Board Cleaner	Sanford	81800	Liquid	Bellwood	ILL		Plastic	1	128 oz	
	Lysol All Purpose Cleaner	Lysol		Liquid	Parsippany	NJ		Plastic	1	6 oz	Dated to remove 6/17
	Vinegar	Four Monks		Liquid	Mt. Prospect	ILL		Plastic	1	20 oz	

	Hydrogen Peroxide	Wards		Liquid	Rochester	NY		Glass	1	128 oz	
	Denatured Alcohol	Barr & Co		Liquid	Memphis	TN		Tin	1	3.75 gallon	Flammable
	Hydrogen Peroxide 3%	Rite Aid		Liquid	Harrisburg	PA		Plastic	1	16 oz	
	Sodium Hydroxide	Science Kit & Boreal		Liquid	San Opispo	CA		Plastic	1	10 ml	
	Biuret Reagent	Science Kit & Boreal		Liquid	San Opispo	CA		Plastic	1	500 ml	Corrosive
	Buffer Solution	Carolina		Liquid	Burlington	NC		Plastic	1	500 ml	
	Glycerin	Webster McGraw		Liquid				Plastic	1	200 ml	
	Buffer Solution	Carolina		Liquid	Burlington	NC		Plastic	1	500 ml	
	Bromothymol Blue Solution .04%	Fisher Scientific		Liquid	Farlon	NC		Glass	2	200 ml	
	Bromothymol Blue .04%	Carolina		Liquid	Burlington	NC		Glass	1	200 ml	
	Iron Metal filings	Sargent-Welch		Filings	Skokia	ILL		Plastic	1	1 lb	
	Glass Beads	Hubbard Scientific		Beads	Chippewa falls	WI		Plastic	1	2 lb	
	Citric Acid, Monohydrate	Baker		Crystal	Phillipburg	NJ		Plastic	1	1 lb	
	Citric Acid	Delta Ed		Powder	Nashua	NH		Plastic	1	480 g	
	Citric Acid	Natural Science		Crystal				Plastic	1	1 LB	
	Mythol Orange .1%	Post Apple Scientific		Liquid	North East	PA		Plastic	1	200 ml	
	Glycerin	Science Kit & Boreal	89304-02	Liquid	San Opispo	CA		Plastic	1	50 ml	
	Universal Indicator	Cynmar			Yulee	FL					
	Iron filings	Lab Aids		Filings	Farmingdale	NY		Plastic	1	1 oz	
	Pepsin	Science Kit & Boreal		Powder	San Opispo	CA		Plastic	1	25 g	

102	AmQuel Plus	Kordon		Liquid	Hayward	CA		Plastic	1	4 oz	
102	Chlor Out	Wardley		Liquid	Secaucus	NJ		Plastic	1	4 oz	
Storage 6	Calcium Fluoride	Bell Telephone Labs Kit		Powder	New York	NY		Plastic	16	1 gr	
	Sodium Hypophosphite	Bell Telephone Lab Kit		Powder	New York	NY			16	1 gr	
	Boric Acid	Bell Telephone Lab kit		Powder	New York	NY			16	1 gr	
	Carborundum	Bell Telephone Lab Kit		Powder	New York	NY			16	1 gr	
	Ammonium Citrate	Bell Telephone Lab Kit		Powder	New York	NY			16	1 gr	
	Fischer Technical Pump Oil	Fischer		Liquid	Schaumburg	IL		Plastic	1		
	Dow Vacuum Grease	Dow Corning		Gel	Midland	MI		Metal tube	1	150 g	
	Premium Pump Oil	Robinar		Oil	Montpelier	OH		Plastic	1	4 oz	
	Sting Ray in unknown liquid/Pan Pacific will test										

In compliance with the California Department of Education's Science Safety Handbook, 2014

009869 = Corrosive Lockable Cabinet in Locked closet between science lab rooms 25 and 28

009870 = Flammable Lockable container in Science Lab room 25

WC = Wood cabinets in Locked Closet between Science Lab rooms 25 and 28

DATE These chemicals will be retained and are to be dated January 1, 2016 and should be removed no later than June, 2017